

Mineral Physics I

Chapter 4. Equation of State

Section 6. Thermal equation of state

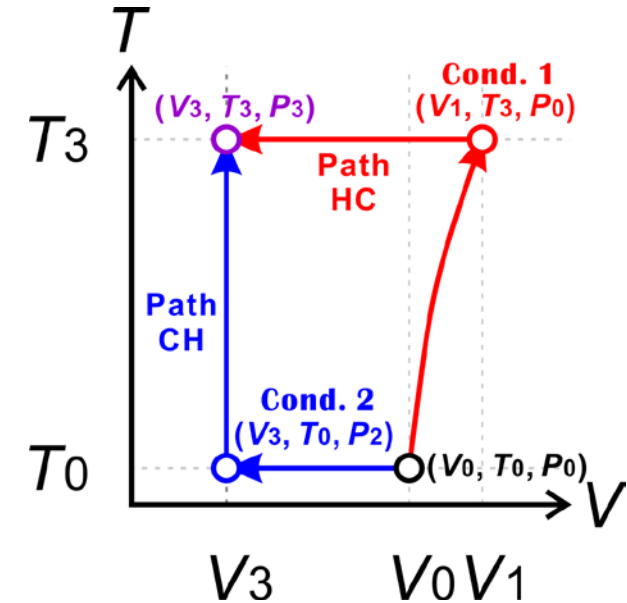
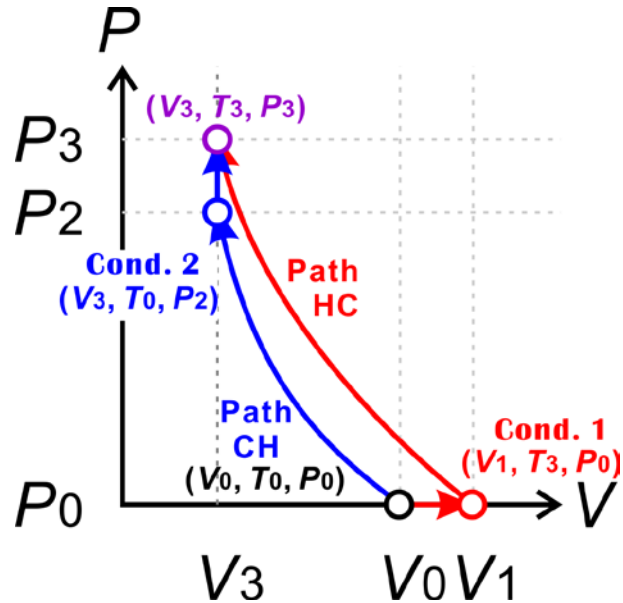
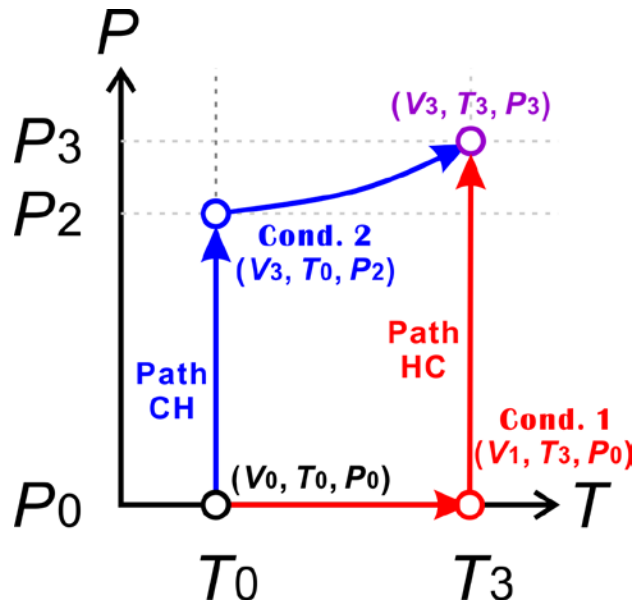


Thermal equation of state

- Thermal equation of state: an equation of state including T as well as P .
- Thermal equation of state is essential for geophysics, because the planetary interiors are under high temperature conditions
 - ✓ Top of the asthenosphere: 1700 K
 - ✓ 660-km discontinuity: 2000 K
 - ✓ Core-mantle boundary: 4000~5000 K



Paths HC and CH



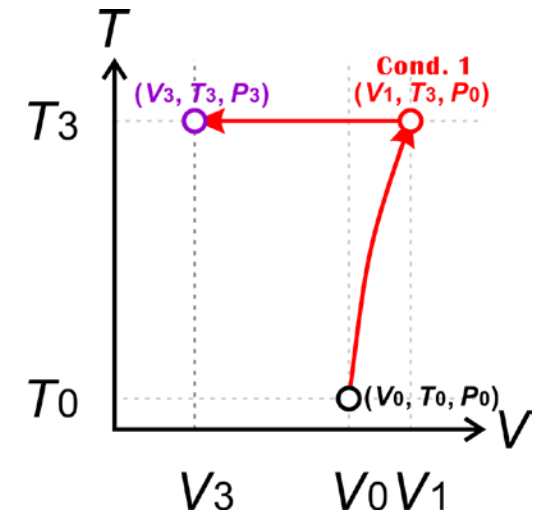
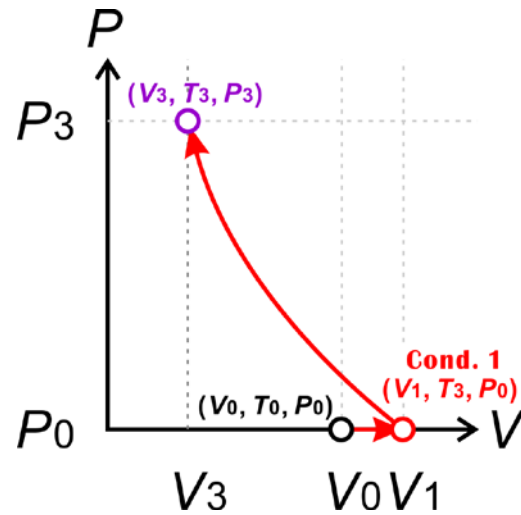
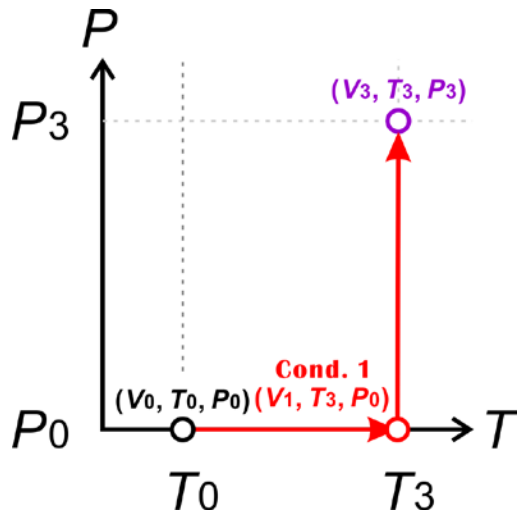
➤ Two paths to reach high P-T conditions (P_3, V_3, T_3) from ambient conditions of (P_0, V_0, T_0)

- ✓ **HC path**: first heated then compressed
- ✓ **CH path**: first compressed then heated



Path HC: HT Birch-Murnaghan EOS

-1



➤ V increases from V_0 to V_1 by heating from T_0 to T_3 at the initial pressure P_0 due to thermal expansion.

$$\checkmark V_1 = V_0 \int_{T_0}^{T_3} \alpha(T)$$

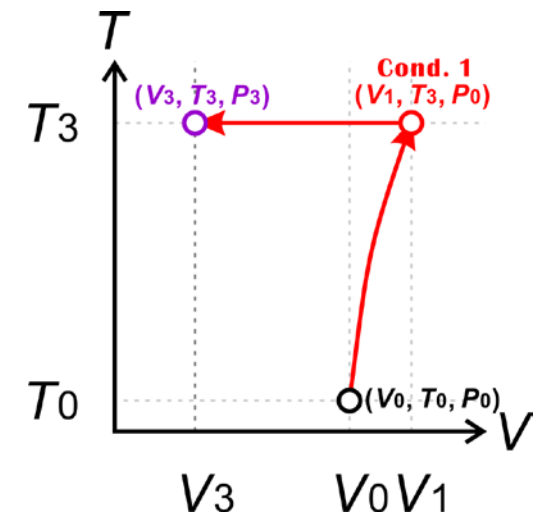
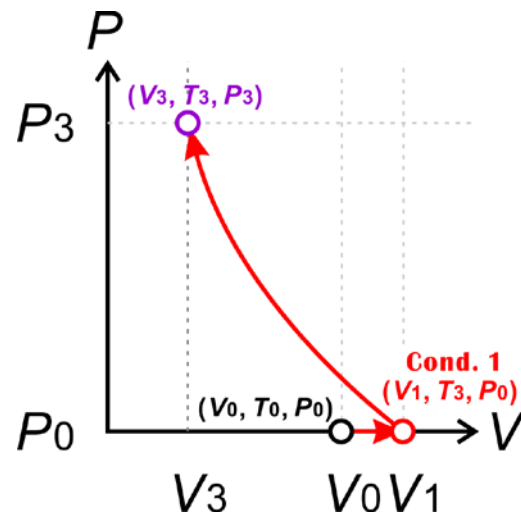
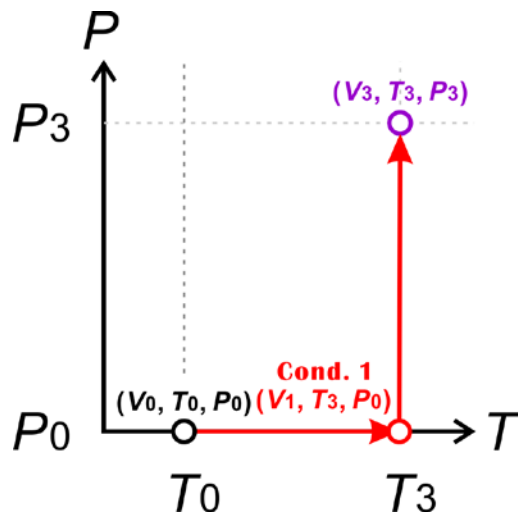
(4.6.1)

- α : thermal expansion coefficient



Path HC: HT Birch-Murnaghan EOS

-2



➤ Assuming the constant thermal expansion coefficient α_0 , we have

$$\checkmark V_1 \cong \{1 + \alpha_0(T_3 - T_0)\}V_0 \quad (4.6.2)$$

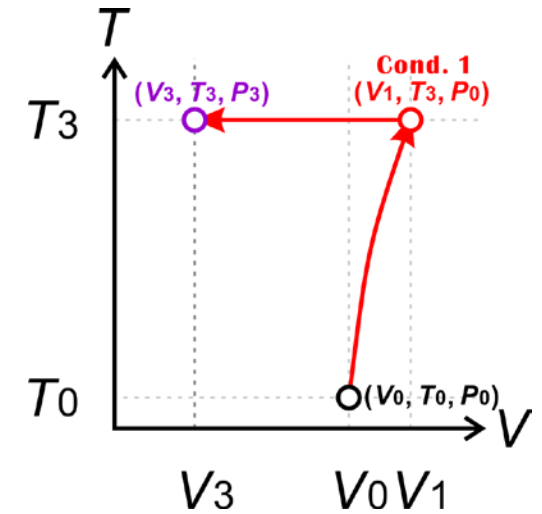
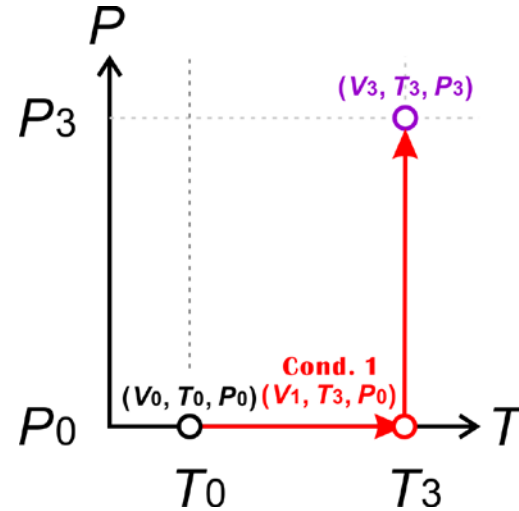
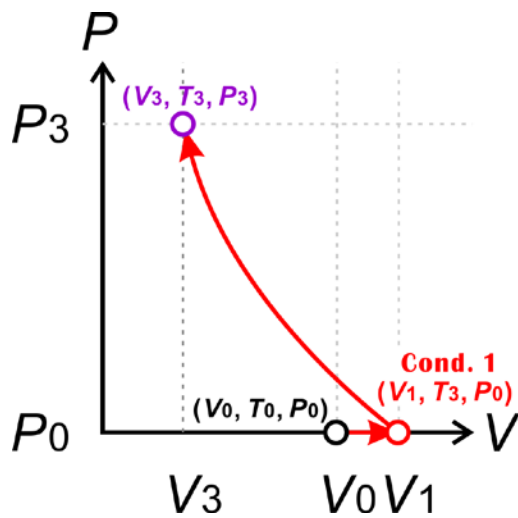
➤ By assuming that thermal expansion coefficient is a linear function of T , we have

$$\checkmark V_1 \cong \left\{1 + \alpha_0(T_3 - T_0) + \frac{1}{2}\alpha_1(T_3 - T_0)^2\right\}V_0 \quad (4.6.3)$$



Path HC: HT Birch-Murnaghan EOS

-3



- Compression from V_1 to V_3 to increase P from P_0 to P_3 expressed by high-temperature 3rd-order Birch-Murnaghan EOS

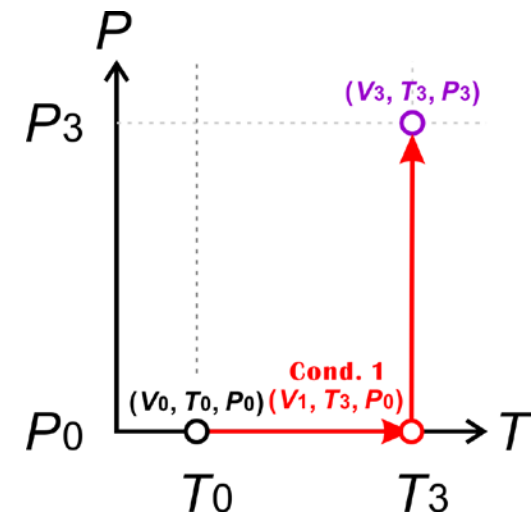
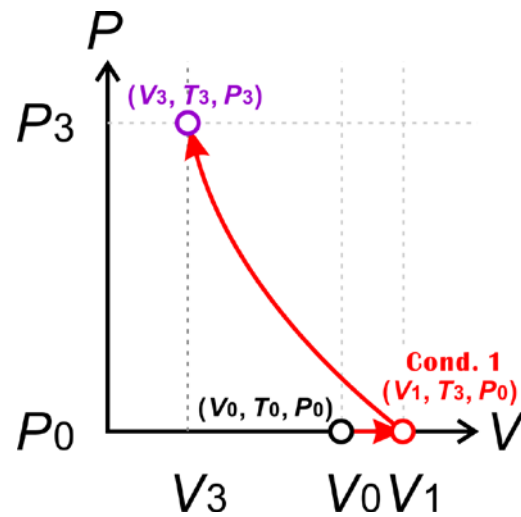
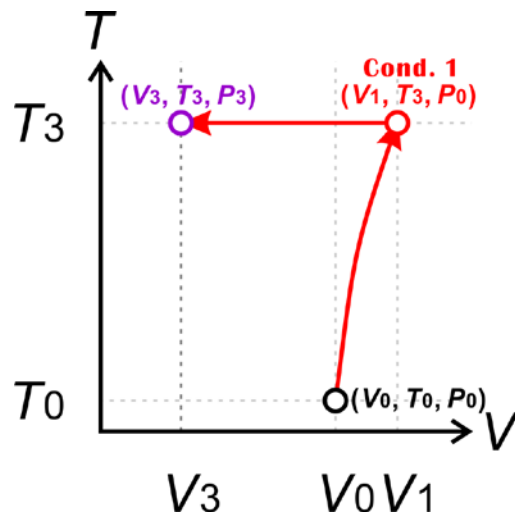
$$\checkmark P_3 - P_0 = \frac{3}{2} K_{T,0}(T_3) \left[\left(\frac{V_1}{V_3} \right)^{\frac{7}{3}} - \left(\frac{V_1}{V_3} \right)^{\frac{5}{3}} \right] \times \left\{ 1 + \frac{3}{4} (K'_{T,0} - 4) \left[\left(\frac{V_1}{V_3} \right)^{\frac{2}{3}} - 1 \right] \right\} \quad (4.6.4)$$

- $K_{T,0}(T)$ is the isothermal bulk modulus at zero P as a function of T



Path HC: HT Birch-Murnaghan EOS

-4



➤ For simplicity, $K_{T,0}(T)$ is assumed a linear function of T

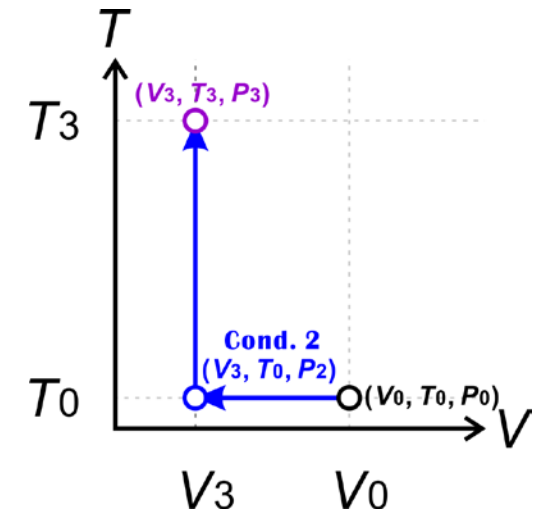
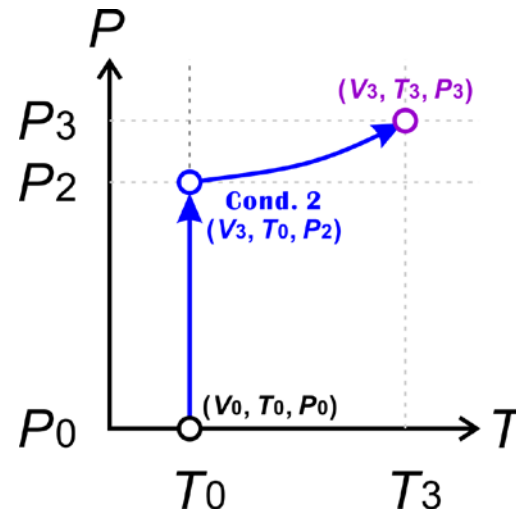
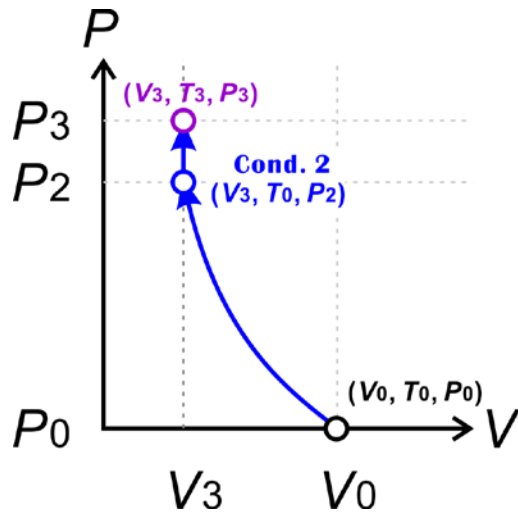
$$\checkmark K_{T,0}(T_3) \cong K_{T=T_0,0} + \left(\frac{\partial K_{T,0}}{\partial T} \right)_P (T_3 - T_0) \quad (4.6.5)$$

Because of experimental difficulty, $K'_{T,0} = \left(\frac{\partial K_{T,0}}{\partial P} \right)_T$ is usually assumed independent from P and T



Path CH: Mie-Grüneisen equation of state

-1

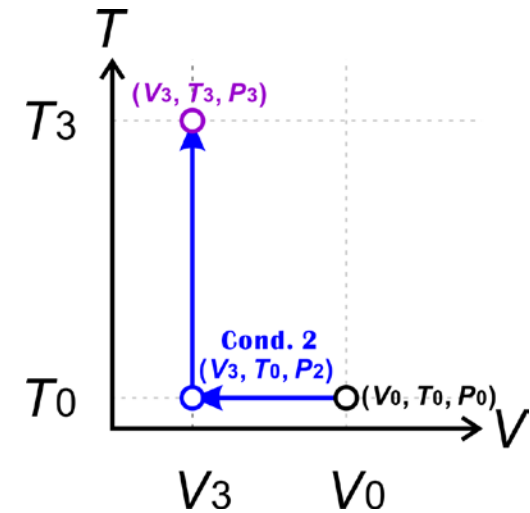
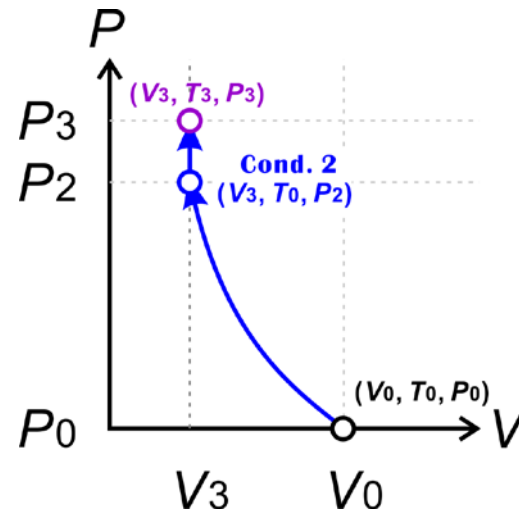
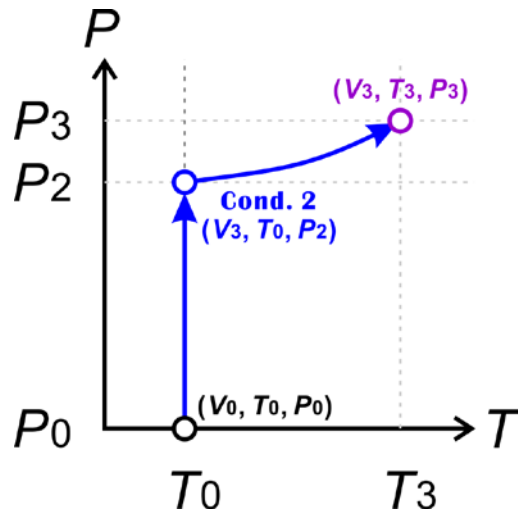


➤ V is first decreased from V_0 to V_3 at $T = T_0$ to increase P from P_0 to P_2

$$\checkmark P_2 - P_0 = \frac{3}{2} K_{T,0}(T_0) \left[\left(\frac{V_0}{V_3} \right)^{\frac{7}{3}} - \left(\frac{V_0}{V_3} \right)^{\frac{5}{3}} \right] \times \left\{ 1 + \frac{3}{4} (K'_{T,0} - 4) \left[\left(\frac{V_0}{V_3} \right)^{\frac{2}{3}} - 1 \right] \right\} \quad (4.6.6)$$



Path CH: Mie-Grüneisen-EOS -2



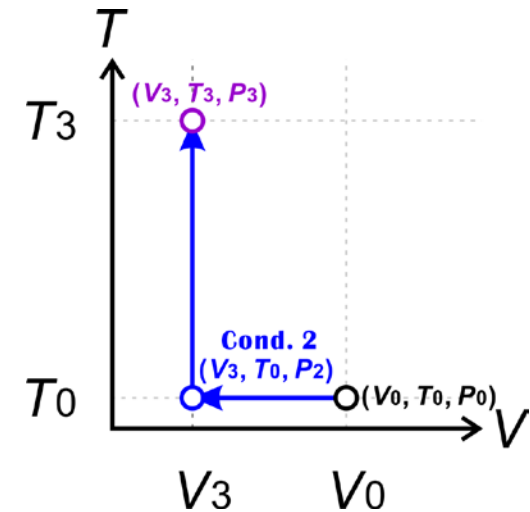
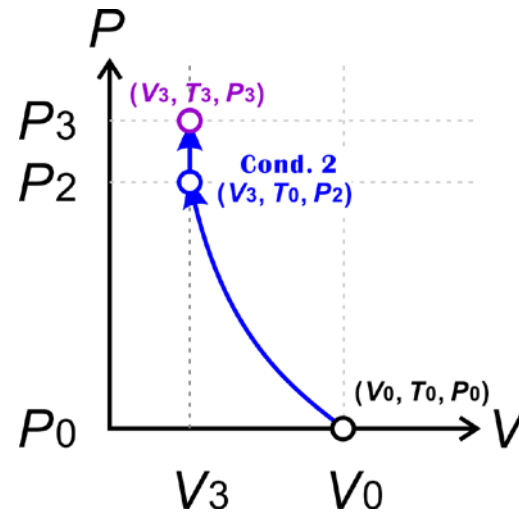
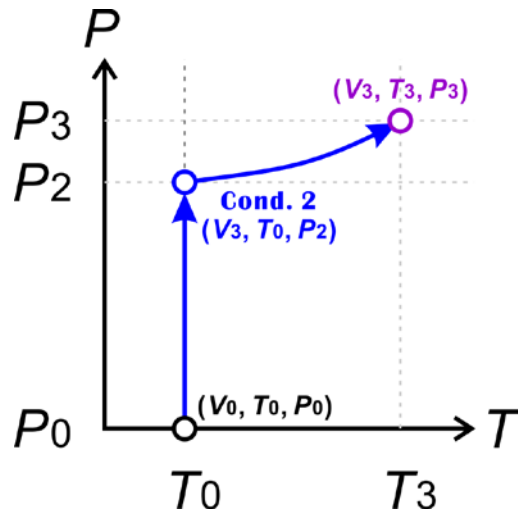
➤ P increase $\Delta P_{\text{th}} = P_3 - P_2$ by T increase at constant volume $V = V_3$

$$\checkmark \Delta P_{\text{th}} = P_3 - P_2 = \int_{T_0}^{T_3} \left(\frac{\partial P}{\partial T} \right)_{V=V_3} dT \quad (4.6.7)$$

- $(\partial P / \partial T)_V$: thermal pressure



Path CH: Mie-Grüneisen-EOS -3



➤ ΔP_{th} : related to E increase by T increase at constant V

$$\checkmark \Delta P_{th} = \gamma_{th} (\Delta E_{th} / V)$$

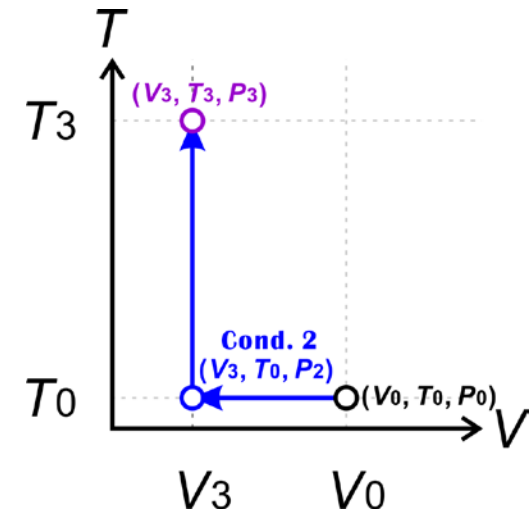
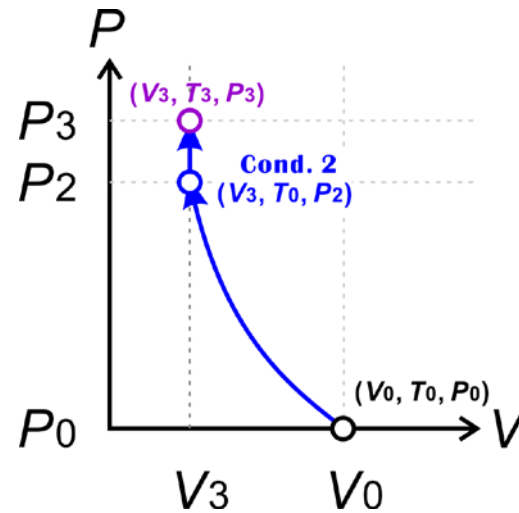
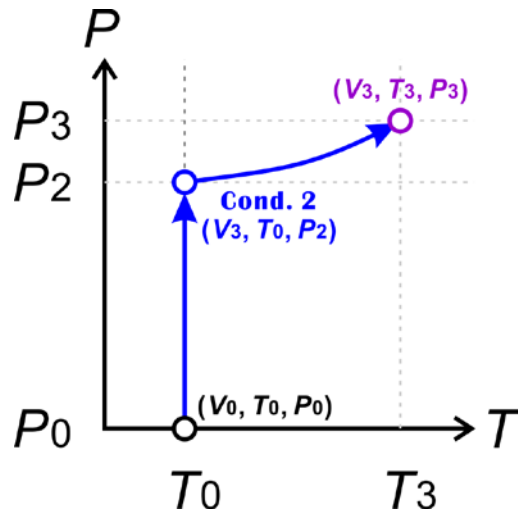
(4.6.8)

- γ_{th} : Grüneisen parameter
 - Definition of γ

- Eq. (4.6.8): **Mie-Grüneisen equation of state**



Path CH: Mie-Grüneisen-EOS -4



➤ ΔE_{th} is expressed by isochoric heat capacity C_V :

$$\checkmark \Delta E_{\text{th}} = \int_{T_0}^{T_3} C_V dT$$

(4.6.9)

➤ By the Debye approximation,

$$\checkmark C_V = 9Nk_B (T/\theta_D)^3 \int_{\theta_D/T_0}^{\theta_D/T_3} \{x^4 \exp x / (\exp x - 1)^2\} dx$$

(3.7.18')



Path HC or CH?

- Which is better, Path **HC** or **CH**?
 - ✓ Although some one says that the high-temperature Birch-Murnaghan EOS does not have physical meaning, but the Mie-Grüneisen-Debye EOS does, however, ...?
- Merits of **CH path**
 - ✓ Physical properties under HT and RP conditions for the **HC path** are more difficult to obtain those under HP and RT conditions for the **CH path**
 - ✓ The thermal pressure for the **CH path** is more easily estimated due to the Debye model than high-temperature compression for the **HC path**



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End

